LORD'S ADDITIVES LLC

856-415-1374 856-415-0227 (F))

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то: Matthew Daniels	FROM: Stephen Bales
COMPANY: USPTO	DATE: 3/9/2008
FAX NUMBER: 571-273-2450	TOTAL NO. OF PAGES INCLUDING COVER:
PHONE NUMBER: 571-272-2450	sender's rephrence number.
RM:	YOUR PERENCHANCE ANIMORS.

Mr. Daniels.

We have completed our response to the BPAI decision of Jan 24th - a draft is Attached.

We would like an interview regarding this draft, but realize that your schedule may rule this out since our response is by before March 24th.

Our amendment is in three parts

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- a. The invention obtained unexpected results
- b. Amended claims 10 & 15 and new claim 13 eliminate elements found in prior art
- c. Two boron chemicals taught in Lloyd (colemanite & ulexite) would not function satisfactorily in Aida, making Aida's teaching not satisfactory for purpose.

If your schedule wont permit review of the entire package, but would allow review of items b and c (pages 9 and 10) this would be helpful to us.

Regards, Stephen Bales Mar 09 08 08:14p Stephen & Sandra Bales

Sandra Bales 856-415-0227

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IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

Appl. No.

10/681,497

Applicant

Stephen G. Bales

Filing Date

October 8, 2003

Title

Lignocellulosic, Borate Filled, Thermoplastic Composites

Examiner:

Matthew I Daniels

Art Unit

1732

: LA 001

Docket No.
Customer No.

000048373

February 28, 2008

Commissioner for Patents P.O. Box 1450

Alexandria, VA 22313-1450

Sir:

In response to the Board of Patent Appeals and Interferences decision of 1/24/2008, Applicant requests to reopen prosecution under 37 C.F.R § 41.50(b) and submits the following amendment to application 10/681.497.

AMENDMENT

Amended, Deleted and New Claims begin on page 2 of this paper.

Remarks/Arguments begin on page 3 of this paper.

A Complete "Clean Sheet" Set of Claims is presented in Appendix A

New Evidence described in the Remarks section is contained in the attached IDS #5.

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Amended, Deleted, and New Claims

- (Currently Amended) In the method for forming lignocellulosic thermoplastic
 composite products such as to increase their resistance to surface visual impairment
 caused by mold attack, the improvement which comprises incorporating an amount of
 a boron-containing fungicide <u>in the range of from about 2 to 12 percent</u>
 by weight of said composite product prior to forming said composite product.
- 2 Deleted
- (Currently Amended) The method according to claim 8 in which said calcium borate is selected from the group consisting of nobleite, gowerite, hydroboracite, ulexite, and colemanite.
- (Currently Amended) The method according to claim 6 in which said calcium borate is a synthetic-borate. 1 in which said boron-containing fungicide is boric acid.
- 13. (New) In the method for forming composite products consisting of a thermoplastic material, a lignocellulosic material, and at least one of the group consisting of a lubricant, a cross-linking agent, a process aid, a UV stabilizer, a blowing agent, an inhibitor, and a coupling agent, such as to increase their resistance to surface visual impairment caused by mold attack, the improvement which consists of incorporating an amount of a boron-containing fungicide selected from the group of zinc borate, synthetic calcium borate, colemanite, ulexite, boric acid, or mixtures thereof in the range of from about 2 to 12 percent by weight of said composite product prior to forming said composite product.
- (Currently Amended) The method according to claim 1 in which said lignocellulosic thermoplastic material is wood polyvinyl chloride.

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Remarks/Arguments

The Applicant requests the claims following amendment be accepted. The arguments are presented in three parts:

- (a) The invention produced unexpected results.
- (b) Amended claims 10, and 15 and new claim 13 eliminate elements contained in the prior art (Aida).
- (c) The use of colemanite or ulexite in Aida would not provide satisfactory flame retardancy, making Aida's teaching unsatisfactory for its intended purpose.

A. THE PRESENT INVENTION PRODUCED UNEXPECTED RESULTS

The present invention revealed the surprising discovery that only when the loading of a boron-containing fungicide into a lignocellulosic thermoplastic composite reaches the 2 percent level does the fungicide become an effective mold resistant agent. And the fact this resistance is maintained for a long duration under adverse outdoor conditions is also surprising when compared with prior art.

Invention Timeline and Overview

A Provisional Patent Application (PPA) was filed in Nov. 2002 and the Patent

Application 10/681,497 filed on Oct. 8, 2003. Example 1 of the application contained

mold data collected in an 18 month outdoor exposure.

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Prior Art: Zinc Borate Thought by Experts to Be Ineffective in Providing Lignocellulosic Thermoplastic Mold Resistance for Years After This Invention

As of May 2005 there were two public studies regarding the use of boron-containing fungicides to reduce surface impairment due to mold in lignocellulosic thermoplastics.

Neither was published at the time the PPA was filed.

On May 20, 2003 Dylingowski presented Maintaining the Aesthetic Quality of WPC

Decking with Isothiazolone Biocide at the "7th International Conference on WoodfiberPlastic Composites" sponsored by the Forest Products Society. This study compared the
effectiveness of two fungicides: Dichloro-Octyl-Isothiazolone (DCOIT-p16) and Zinc

Borate (p(19) in reducing mold growth in wood-plastic composites. Dylingowski's

presentation contains the following key points:

- a. The list of molds and fungi that can colonize on wood (p3) is not identical to those that can colonize on plastics (p4,7). This implies that a fungicide that is effective on wood may not be effective on plastic and vice-versa. As a result, one skilled in the art would conclude that if a chemical is ineffective in reducing mold in lignocellulosic or thermoplastic composites it would not work in lignocellulosic thermoplastics. Conversely, effectiveness against mold in those two composites would not guarantee the chemical will be effective in lienocellulosic thermoplastics.
- b. Zinc borate loaded at 1 percent was not effective in resisting 5 types of mold fungi under standard laboratory test ASTM G-21-96 (p 25 to 30), was not effective in resisting a culture of Aspergillus (p 31 to 35) and had slight success against 5 types of fungi using MIL.STD.810E testing (p36-39).

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Dylingowski concludes that "Zinc Borate did not prevent mold growth in

laboratory tests" (p40).

Dylingowski, one skilled in the art of mold growth on wood, plastics, and wood-

Plastics, concluded that zinc borate is not effective in restraining mold growth on

wood plastics. Further, the data presented on pages 27 & 32 indicates the lack of a dose-

response relationship; this leads one skilled in the art to the conclusion that the borate

would have no effect even at increased doses. The present invention's discovery that at 2

percent or greater zinc borate does restrain mold growth is an unexpected result. And the

fact that it does it for a long period of time under adverse outdoor exposure conditions is

even more surprising.

On May 23, 2005, Laks et al published Effect of Manufacturing Variables on Mold

Susceptibility of Wood-Plastic Composites (copy submitted in Applicants IDS #2). This

material was also presented at the "8th International Conference on Woodfiber-Plastic

Composites" sponsored by the Forest Products Society.

Laks validates the results of the present invention, which reveals that the use of zinc

borate is effective at reducing mold at the 3 and 5 percent levels (p10). He demonstrates

that a surprising change occurs in surface mold's dose-effect response to zinc borate

starts between 1 and 3 percent, something demonstrated by this present invention years

earlier.

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Laks also noted evidence indicated that mold growth was a significant problem on wood plastic decking installations. With 83 publications and 4 patents in the field of wood and wood plastic preservation, the Applicant considers Dr. Laks to be one

exceptionally skilled in the art.

Laks states mold growth was a significant problem for the industry, but he and others

skilled in the art did not realize until years after the present invention that zinc borate

could reduce wood plastic surface mold growth, and then only at 2 percent higher

loadings. This shows that Applicant's discovery was not obvious.

Prior Art: Colemanite Ineffective in Providing Thermoplastic Mold Resistance

As previously noted, the lack of a fungicide's mold resistance effectiveness in thermoplastics is a good indicator that it will not work well in lignocellulosic

thermoplastics.

Koskiniemi's teaching regarding the use of calcium borate in thermoplastics was

available prior to the PPA filing. It indicated synthetic calcium borate had to be modified

to provide long term mold protection and that colemanite, the naturally occurring form of

calcium borate, did not provide this protection even in the short term.

Koskiniemi (USPN 5,482,989) at Example 1, Tables 2 and 3, (4:20 - 5:29) teaches

that the lack of resistance to mold growth of thermoplastic samples loaded with 2 percent

colemanite (Sample D) was similar to a sample loaded with 2 percent talc (Sample C).

This result is particularly remarkable since talc has no fungicidal properties.

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Modified calcium pyroborate (Sample A) did provide mold resistance. However the modification required coating with silica to reduce its solubility. Koskiniemi teaches that such coating would be required for plastics that would be placed in use over along time period (2:44-59). But the current invention is designed for wood plastic that will be used on decking and railing for many years.

Koskiniemi tested samples using molds from the Aspergillus family, which Dylingowski reveals (p8) is one of the molds that can colonize on wood plastic surfaces. Koskiniemi teaches that indicates colemanite would not be effective at 2 percent loading and further states that without a silica coating synthetic calcium borate would not be effective in the long run. However the current invention discovered that colemanite and non-coated synthetic calcium borate is effective in both the short and long term. Example 1 demonstrated that colemanite was effective in an 18 month outdoor exposure.

Prior Art: Zinc Borate Ineffective In Providing Lignocellulosic Mold Resistance

The prior art that existed at the time of the application filing, and even that published at a later date, teaches that zinc borate is not an effective against mold fing.

Marks, et al (USPN 6,416,789) teaches "...it is known to use zinc borate to protect cellulosic composites, including particleboard, hardboard, and oriented strand board, from fungal decay as described in US Patent Nos. 4,879,083, 5,763,338, and 5,972,266. Zinc borate is usually applied as a powder or a liquid suspension to wood chips or strands, prior to their fabrication into panels. However, zinc borate is not a very cost effective wood preservative, and does not provide good protection against mold, mildew and staining fungi at typical use levels" (1:24-35).

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Marks does not provide typical use levels, but recommended use ranges are identified in his referenced patents and also in an EPA document. USPN 5,972,266 (Fookes et al) teaches "...5 to 25 gm of zinc borate per kg of wood strands ..." (1:50-55) or 0.5 to 2.5 percent by weight of zinc borate. USPN 5,763,338 (Sean) at 2:37-49 presents an invention that allows the incorporation of zinc borate in lignocellulosic composites at levels greater than 1 percent.

Finally, the Environmental Protection Agency (EPA) requires the formal registration of any chemical used as a fungicide. As part of this registration process, an applicant must submit a label and an associated technical data sheet describing how the fungicide is to be used. The technical data sheet submitted to the EPA describing the use of zinc borate as a fungicide in wood composites identifies that the loading must be at least 0.5 percent and must not exceed 8 percent depending on the wood species used.

Ross (USPN 7,056,919) teaches "For example, it is known to use zinc borate to protect cellulosic composites, including particleboard, hardboard and oriented strand board, from fungal decay as described in U.S. Pat Nos. 4,879,083; 5,763,338; and 5,972,266. However, like most borates, zinc borate is not particularly effective against mold fungi" (1: 46-52).

Prior Art: Zinc Borate is Ineffectivein Providing Coating Mold Resistance Batdorf (USPN 6,881,247) teaches the addition of a combination of a metal borate, zinc oxide, and magnesium hydroxide to provide an effective antifungal coating additive. His definition of a metal borate includes zinc borate and calcium borate (3:27-29).

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Batdorf them teaches "metal borates by themselves are not very effective for inhibiting growth over a broad specimen of fungi strains or against many bacteria. However, when used along with magnesium hydroxide, a synergism is obtained resulting in broad and effective protection (3: 62-67)."

B. AMENDED CLAIMS 10 AND 15 AND NEW CLAIM 13 REMOVE ELEMENTS EXISTING IN THE PRIOR ART

Claim 10 has been narrowed to only include boric acid, a boron-containing fungicide that is covered in the Specification but not taught in Aida (USPN 5,221,781). Similarly, claim 15 has been narrowed to include only polyvinyl chloride, a thermoplastic not taught in Aida.

New claim 13 eliminates the inclusion of a fire retardant other than zinc borate, synthetic calcium borate, colemanite, ulexite, or boric acid. The Examiner agreed that Aida teaches that if only inorganic flame retardants are used they should be used in combination with organic flame retardants (April 18,2007 Answer, p9). Claim 13 eliminates organic flame retardants.

C. USE OF COLEMANITE AND ULEXITE AT THE 2 TO 12 PERCENT LEVELS OF THE PRESENT INVENTION WOULD RENDER AIDA UNSATISFACTORY FOR ITS INDENDED PHIRPOSE

Claim 9 identifies colemanite as the boron-containing fungicide. The BPAI agreed with the Examiner that Aida teaches the amount of flame retardants, such as zinc borate, could be present in an amount of 5 to 200 parts by weight of the resin. As a result they agreed Aida provided a fair teaching of employing the boron-containing compound in the claimed amount of about 2 to 12% by weight of the composite product.

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However, Bosserman (USPN 6,723,352) and Pitts (USPN 3,865,760) teach that colemanite and ulexite would not effectively serve as a flame retardant in those ranges.

rendering Aida's invention unsatisfactory for its intended purpose if colemanite or ulexite

is added in the 2-12 percent range. Bosserman states ""calcium borate ores have been

previously used as components in dry powder flame-retardant formulations. One such

use was described in U.S. Pat No 3,865,760 to Pitts, et al., wherein the ore colemanite (or

alternatively, the ores ulexite or pandermite) was used as a filler in a rubber and plastic

dry powder formulation, alone or in combination with alumina trihydrate and calcium

carbonate. In this formulation, high levels of unreacted dry colemanite were required in

order to receive the desired flame-retardant effect" (1:64 - 2:6). Pitts Table 1 shows that when colemanite is used alone, a loading of 150 parts by weight of the dry rubber latex is

required. At the lowest thermoplastic loading of 25 percent noted in the Specification,

12 me to west attended to the paster to the specification,

this would require a 37.5 percent loading of colemanite or ulexite.

Summary

For all reasons given above, Applicant respectfully submits the distinctions are of

patentable merit under Section 103. Accordingly Applicant submits this application is

now in full condition for allowance,

Respectfully Submitted,

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Appendix A

- (Currently Amended) In the method for forming lignocellulosic thermoplastic
 composite products such as to increase their resistance to surface visual impairment
 caused by mold attack, the improvement which comprises incorporating an amount of
 a boron-containing fungicide in the range of from about 2 to 12 percent
 by weight of said composite product prior to forming said composite product.
- 2. (canceled)
- The method according to claim 1 in which said amount of boron-containing fungicide is in the range of from about 3 to about 5 percent by weight of said composite.
- 4. The method according to claim 1 in which said lignocellulosic material is selected from the group consisting of wood, ground rice hulls, kenaf, jute, and coconut shells.
- The method according to claim 1 in which said thermoplastic material is selected from the group consisting of polyethylene, high-density polyethylene, polystyrene, and polyvinyl chloride.
- The method according to claim 1 in which said boron-containing fungicide is calcium horste
- 7. (canceled)
- The method according to claim 6 in which said calcium borate is a naturally occurring borate.
- 9. (Currently Amended) The method according to claim 8 in which said calcium borate

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is selected from the group consisting of nobleite, gowerite, hydroboracite, ulexite and colemanite.

- (Currently Amended) The method according to claim 6-in which said-ealeium borate
 -is-a synthetic-borate. 1 in which said boron-containing fungicide is boric acid.
- 11. The method according to claim 1 in which said boron-containing fungicide is selected from a group consisting of zinc borate, calcium borate, boric acid, or mixtures thereof
- 12. The method according to claim 8 in which said calcium borate is colemanite.
- 13. (New) In the method for forming composite products consisting of a thermoplastic material, a lignocellulosic material, and at least one of the group consisting of a lubricant, a cross-linking agent, a process aid, a UV stabilizer, a blowing agent, an inhibitor, and a coupling agent, such as to increase their resistance to surface visual impairment caused by mold attack, the improvement which-consists of incorporating an amount of a boron-containing fungicide selected from the group of zinc borate, synthetic calcium borate, colemanite, ulexite, boric acid, or mixtures thereof in the range of from about 2 to 12 percent by weight of said composite product prior to forming said composite product.
- 14. The method according to claim 1 in which said boron-containing fungicide is zinc borate
- (Currently Amended) The method according to claim I in which said lignocellulesie thermoplastic material is wood polyvinyl chloride.